

# Chemistry 1C, General Chemistry and Quantitative Analysis

## Chem 1C - Section 01Y

Lecture MW – **Synchronous Online** - 4:30PM to 5:45PM

Lab MW- **In-Person** – 11:30AM to 2:20PM – SC2208

## Chem 1C - Section 02Y

Lecture MW – **Synchronous Online** - 4:30PM to 5:45PM

Lab TR- **In-Person** – 11:30AM to 2:20PM – SC2208

**Instructor:** Dr. Chris Deming, email: [demingchristopher@fhda.edu](mailto:demingchristopher@fhda.edu)

**Office Hours:** Office hours will be held through zoom. The days/times are Mondays 5:45 PM – 7:00 PM, Tuesdays 4:00 PM – 5:30 PM, and Wednesdays 5:45 PM – 7:00 PM. The room is the same as the one for lectures and the link is given on page 5.

**Course Description:** This class will cover the principles of solutions, buffers, electrochemistry, transition metals, and nuclear chemistry.

This course is divided into two separate instructional periods, the lecture and laboratory sections. The lecture portion is primarily devoted to the material discussion while the laboratory portion gives a chance for chemical experimentation. One registration code will enroll for the lecture and lab sections. Lecture and lab sections must be taken together to pass Chem 1C and will both go towards a single grade.

## **Course Material:**

**1. Primary Lecture Text:** CHEMISTRY: The Molecular Nature of Matter and Change, Silberberg and Amateis, 9e. Other editions will be essentially the same and will work great to study, but practice problems given in this syllabus correspond to the 9<sup>th</sup> edition. Although there are online text options, we will NOT be using CONNECT or ALEKS this quarter. There are multiple options to obtain the text for this course depending on your specific needs.

- a. **Option 1** – Hard copy text. This can be used and any edition. I will give homework and solutions from the 9<sup>th</sup> edition, but you will find these questions in any edition but potentially with different numbering. This is the way to go if you prefer hard copies and is a great choice if you want a quality chemistry textbook to reference in the future.
- b. **Option 2** – 90 day access to an electronic text specifically for CHEM 1C. This is a great, cheap option that will give you 90 day access to an electronic text

for the chapters in this class (CHEM 1C) only. This can be purchased with the ISBN: 9781307600971.

**2. Supplementary Lecture Text:** Calculations in Chemistry. While homework problems will not be assigned from this book, more practice is always a good idea, and this book can give just that. It can be found as a hard copy or digital through the De Anza bookstore.

**3. Lab Equipment:** This quarter, we will continue labs in person. Goggles and masks are required for the entire lab session and will be provided in the lab, so all that is left for you to bring is the appropriate clothing for lab work, such as long pants and closed toe shoes, as well as a lab notebook. This notebook cannot be pocket size and must be permanently bound. Other than that, the type doesn't matter.

**4. Scientific Calculator.** Logarithm and exponential functions required. You are encouraged to bring your calculator each day to work through examples as they are presented. Phones will not be allowed for calculations during tests.

**5. Camera linked to the internet** – For much of your classwork, you will need to take a picture of your work and submit to CANVAS. Phones are 100% okay. Please let me know if this is an issue as soon as possible.

**Class Registration.** Registration limit is strictly set at 30 per section since we limited by the space in lab. The class will be filled based on the official roster provided by the De Anza Admissions and Records, including an official waitlist. Students on this waitlist may attend the online lecture within the first two weeks but will not be allowed to come to lab until officially enrolled due to space restrictions. Since lecture meetings will be through Zoom, I will create a waitlist mailing list to give invites to lecture and send assignments up until the registration is finalized at the end of the second week.

**Resources:** Academic support can be found at the Learning Resources Division <https://www.deanza.edu/learningresources/>. Information about tutoring can be found at the Math Science and Technology Resource Center <https://www.deanza.edu/studentuccess/mstrc/>.

**Academic Integrity:** By enrolling in classes at De Anza College, you are agreeing to the academic integrity policy and are held to all standards. Specifics can be found at <https://www.deanza.edu/studenthandbook/academic-integrity.html>.

Cheating during an exam/quiz or copying/using work other than your own for assignments will result in a 0 for the entire assignment, regardless of what percentage of the work is from cheating.

Worse than a 0 on an exam, I am required to report such incidents to the disciplinary committee, who will make a note of the incident on your transcript, which then becomes visible to 4 year colleges upon reviewing your transfer application.

**Disability Service Support:** De Anza is committed to providing support for all students. Please contact me as soon as possible if you require special accommodations and I will be happy to do what I can to help. For more information, visit Disability Service Support at <https://www.deanza.edu/dss/>

**Classroom Conduct:** I want to be very clear that this class is a place where everyone can feel safe to be themselves and to learn at their own pace. It is important to me that you feel comfortable to ask questions, and I hope you all will help me create a supportive atmosphere.

**Course Schedule.** All exam dates, lecture topics/dates, lab topics/dates are listed on page 10. The final exam date is also provided on page 10 and can also be found on the de Anza website.

**Grade Assignment.** This rubric is subject to change throughout the quarter.

Grade	Percentage
A+	>97
A	97-93
A-	93-90
B+	90-87
B	87-83
B-	83-80
C+	80-76
C	76-70
D	70-60
F	<60

## Class Assignments:

Assignment	Points	Percent
Student Welcome Questionnaire	5.0	0.6
CHEM 1B Review Sheet	20.0	2.2
Research Report	40.0	4.4
Green Crystals Revisit	25.0	2.8
Exam 1 - Chapter 13	100.0	11.1
Exam 2 - Chapter 19	100.0	11.1
Exam 3 - Chapter 21	100.0	11.1
Final Exam	150.0	16.7
<b>Lecture Total</b>	<b>540.0</b>	<b>60.0</b>
Chemical Lab Safety Sheet	5.0	0.6
COVID Lab Safety Sheet	5.0	0.6
Freezing Point Prelab	10.0	1.1
Freezing Point Quiz	5.0	0.6
Freezing Point Experiment	5.0	0.6
Freezing Point Calculations	5.0	0.6
Freezing Point Conclusion	10.0	1.1
Buffers Prelab	10.0	1.1
Buffers Quiz	5.0	0.6
Buffers Experiment	5.0	0.6
Buffers Calculations	5.0	0.6
Buffers Conclusion	10.0	1.1
Common Ion Prelab	10.0	1.1
Common Ion Quiz	5.0	0.6
Common Ion Experiment	5.0	0.6
Common Ion Calculations	5.0	0.6
Common Ion Conclusion	10.0	1.1
Anions Prelab	10.0	1.1
Anions Quiz	5.0	0.6
Anions Experiment	5.0	0.6
Anions Calculations	5.0	0.6
Anions Conclusion	10.0	1.1
Electrochemistry Prelab	10.0	1.1
Electrochemistry Quiz	5.0	0.6
Electrochemistry Formal Report	60.0	6.7
Cations Prelab	10.0	1.1
Cations Quiz	5.0	0.6
Cations Experiment	5.0	0.6
Cations Calculations	5.0	0.6
Cations Conclusion	10.0	1.1
Lab Final	100.0	11.1
<b>Lab Total</b>	<b>360.0</b>	<b>40.0</b>
<b>Class Total</b>	<b>900.0</b>	<b>100.0</b>

# Lecture

## Lecture Description

This class (Chem 1C) will cover chapters 13, 19, 21, 23, and 24 from the assigned textbook. All lectures will be held through Zoom during the indicated time period (**synchronous**), so please download this free application. For the lecture and office hours' room, follow the link <https://fhda-edu.zoom.us/j/99281131289?pwd=UkVUdFIVNGljdzZPa01ZWVVhRXlyUT09>

This link will also be on CANVAS under "pages". The power point lecture slides will be posted before the lecture on CANVAS under files, and a link to the recording of the lectures will be posted afterwards under "pages".

My general philosophy toward lecture is that I am looking to give a deep description of the concepts as well as a thorough review of the associated mathematics. The marriage of these two independent ways of understanding is ultimate goal. This means I will spend a significant amount of time in lecture describing the atomic scale dynamics as well as going over calculations.

Below are some helpful tips that make learning much easier this quarter.

**1. Review the material before attending the lecture.** This could include reading the section in the textbook, reviewing the lecture slides, or even glancing at the homework. This will help you develop a stronger and more personal connection the topics and make the presented material much easier to understand.

**2. Don't only copy the words from the slides during lecture.** Since I will give the lecture slides before the lecture, you will not need to copy down everything on the presentation. In fact, many of the slides are very dense with info, and it would take you too much time to copy all the words down.

Furiously copying the words on slides is not the best way to learn during a lecture and instead, writing down what I am saying and/or what you are thinking will be a much more successful method.

This can be done with a tablet, printing out the slides before, or writing in a notebook using the slide numbers.

**3. Complete all homework problems and all of the in-chapter reviews.** Extensive practice is the best way to ensure concept mastery. The more you practice, the more comfortable you will be, and the better you will perform on exams. Beyond the minimum of the assigned problems, you are encouraged to do the in-chapter problems as well as the end of the chapter problems that were not assigned. I am serious when I saw that I sometimes use problems from the book that I didn't assign as exam problems.

**4. Don't fall behind.** In chemistry, each new topic will build on the previous one so it is essential to understand the topics as they are presented. Following a lecture when you do not

understand the previous material is not an effective method for learning and will lead to further problems. To avoid falling behind.....

**5. Get help when you need it.** If you are having a difficult time with a topic, it is your responsibility to get help. There are plenty of resources, including myself, for aiding in material comprehension, but it all starts with you making an effort to get this help. You are also encouraged to find a study group and/or come to office hours.

### Lecture Assignments

**Student Welcome Questionnaire.** In an effort to become better acquainted with you and provide the best overall instructions possible, I have created a set of questions for you to answer, worth **5 points**. My idea is that with a strong relationship and mutual respect, communication and thus learning will be facilitated. My hopes are that knowing some of this info early on will help start building this relationship. The assignment will become available on the first day of class. When you click on that assignment, you will find a pdf with the questions and further instructions.

**CHEM 1B Review Sheet** – To build continuity between chemistry 1B and 1C, and provide a refresher of the previous concepts, there is a chemistry 1B review sheet assignment. This sheet will focus on conceptual and calculation-based questions for the specific parts of chemistry 1B that are expanded upon in 1C. I will talk about this during the first week, and the assignment is worth **20 points**.

**Research Report** – A vital part of scientific work is researching what experiments have been done and using the consequences of those studies to develop a plan to uncover new information through your own experiments. While we won't be doing any original experiments in this class, we will get practice looking for information.

The topic of the report can be anything related to chemistry, but it needs to have some connection to you. While one half of the report is a detailed description of the topic, the other half will be centered on what this topic means to you.

Topics must be approved through email since no topic can be used by more than one student. I will talk about this more and release an instruction sheet during the second week of class. The entire report is worth **40 points**.

**Green Crystals Revisit** – The green crystals review sheet is yet another way we will connect the different classes of the chemistry 1 series and gain more practice with coordination compounds (CH 23). Giving another glance at an old situation with a new lens can be beneficial and will further solidify the specifics of green crystals we did not get in 1B. Once we cover enough information from chapter 23, I will introduce this worksheet, worth **25 points**.

**Lecture Exams.** There will be three lecture exams to test comprehension throughout the quarter. Exams will cover material from lectures, homework, and book chapters. If you are having difficulty completing the homework questions for that chapter, you are urged to get help *before* taking the test. Questions will range from easy to difficult and may require solving

problems that have not been explicitly demonstrated before. I will post a study topic guide a week before the exam.

Each exam is worth **100 points**, and the dates are given in on page 10. No late or early exams will be administered. If you feel the grading of any exam is incorrect, please let me know and I will be happy to talk about talk with me. I will release a key after the exam, and I am very open to hearing what you have to say about the grading, but you must do so within **one week** of the day the exam key is released.

Although the lectures will be delivered remotely this quarter, **the lecture exams will be in-person**, so please make a note of these dates. The room in which the exams will take place will be announced closer to the exam day, but my hopes are FOR1, since it is so large.

**Lecture Final.** The lecture final is worth **150 points** and will cover all chapters, but will have more from chapters 23 and 24, since there will have not yet been any testing on those chapters. The date and time for the final are given on page 10 and will not change

**Homework.** Homework will not be turned in for credit, but will provide the practice needed for concept mastery. The list of homework problems is on page 11. These homework problems will be from the end of the each chapter and will cover calculations as well as conceptually based problems. The solution keys are posted to CANVAS under “files” so you can check your work.

Doing all of the listed problems is highly recommended and represents the minimum needed to practice the topics, but you are strongly encouraged to go beyond the listed problems and try other problems throughout the book. Test questions will be similar to homework questions, so it is important to practice each problem.

# Laboratory

## Lab Description

This quarter, we will be back in the lab again for some in-person experimentation. While this is exciting for many reasons, it is important to be clear about the proper precautions for hazards both old (chemical) and new (COVID), and how to complete lab assignments, since many of us don't have much experience with in-person chemistry labs.

The chemical safety document can be found on the last page of the syllabus. I will go over this sheet in lab and you will read, signed, and turn it in to the CANVAS assignment before performing any experiments worth **5 points**.

Additionally, the chemistry department is committed to providing safe in-person lab work and has agreed upon certain safety protocols to minimize any possible spread of COVID. These protocols are expressed in the COVID lab safety agreement document, which you should already have. Furthermore, there is a self-assessment for students to perform before coming to lab each day. Both the COVID lab safety document and the self-assessment are posted to CANVAS and will be emailed directly to you. You will need to read, sign, and turn the COVID lab safety sheet to the CANVAS assignment for **5 points**. You should go through the

self-assessment each time before coming to class, but do not need to turn it in. We will also discuss these documents more during the first week of lab.

What follows below are descriptions of the policies and assignments required for each experiment.

### Absence Policy

Please go through the self-assessment from the provided document before you go to lab each time. If the result of the assessment is that you should stay home, please do so. While you cannot make up the experiment due to time restrictions, there will be a way for you to make up the points so you won't lose any points for staying home. We are all counting on each other to make the responsible decision and I don't want you coming to lab in fear of losing points when the self-assessment does not clear you to do so.

### Lab Assignments

There will be a total of 6 labs experiments that we will perform this quarter. For each experiment, you are required to read the procedure, complete a prelab, attend the lab introduction at the beginning of the lab period, perform the procedure, answer the follow-up questions/calculations, and write a conclusion. What follows are the steps to complete an experiment and the associated assignments.

Laboratory Step 1: The first thing to do to prepare for the lab is to *read the entire experiment*. It is essential to become familiar with the experimental design and procedures before starting with the lab work, and this starts with a thorough read-through of the methods. The lab documents can be found on the De Anza chemistry website and will also be available as pdfs on CANVAS under "files".

Laboratory Step 2: Once you have familiarized yourself with the lab, the next step is to write a **prelab, worth 10 points**. There are three parts to the prelab that are equally important.

First part of the prelab is the **lab introduction**, where you will describe the goals of the experiment, introduce the scientific principles that form the basis of the study, and summarize the process by which you obtain the experimental data. This should not be a list of procedural steps, but rather 1-3 paragraphs of writing, in your own words.

The second part is a recognition of the **hazards** associated with each chemical in the procedure. This does not need to be everything on the SDS but should convey the hazards working with that chemical and the proper precautions for safe usage. The third part of the prelab is to write **tables** to hold the data you will collect.

All parts of the prelab must be completed before coming to lab. Scan/take pictures of the work and upload them to the appropriate CANVAS assignment before the start time of the lab.

Laboratory Step 3: At the *beginning* of each lab period, there will be a quiz to test how well the procedure has been read and understood. This means there will be a total of 5 lab quizzes worth **5 points** each. These will only take about 10 minutes and you can reference your prelab and lab notebook during the quiz.



Laboratory Step 4: After the quiz, I will give an introduction at the beginning of the lab session that will typically include a discussion of the theory behind the experiment as well as a walkthrough of the harder aspects of the procedure. Missing this time will result in a loss of points and may prevent you from performing the experiment that day.

Laboratory Step 5: After performing the experiment, take a picture of your data (in the pre-made table) and load it to the appropriate assignment before leaving the lab room for **5 points**.

Laboratory Step 6: With the collected data, you will now need to perform calculations and follow up questions, and turn in them into CANVAS for **5 points**. Typically, there will be part of a lab period, or even an entire lab period, dedicated to help with the calculations. The required lab calculations and follow-up questions will be available as a pdf at the beginning of each experiment under the calculation assignment for that lab and are due before attending the next lab session. Due dates will be listed on the canvas assignments.

Laboratory Step 7: The last task is to write a **conclusion**. This section is the most important and often the most difficult because it requires deep consideration of the experiment as a whole. The conclusion should contain at least these three sections.

The first is a summary of the experiment, including the main goal and the methods used to collect/analyze data. This part should not be more than a paragraph and will be very similar to some of the content in the intro.

For the next section, **present the final values**. Many of the experiments require collecting a large amount of raw data, but including all of these values is not the point of this section. Only include the values or conclusions that directly relate to the experimental goal. Additionally, **compare** one trial to the next and/or compare the average value to literature values.

Finally, provide a source of error that may have resulted in discrepancies between trials or between experimental averages and accepted values. This description should go beyond simple factors like human error and should connect an aspect of the experimental design or procedural step to any discrepancies between the experimental and expected values. That is, explain how an error could have affected your result by following this error through the calculation process.

The conclusion will be submitted through CANVAS and is worth **10 points**.

**Formal Laboratory Report.** For the electrochemistry experiment, you are required to do a formal, typed report worth **60 points** that will contain all parts of the lab together in one document, rather than individual parts turned in at separate times, as in the other labs.

In scientific research, conveying what you have discovered in a clear, concise manner is essential to making your new ideas accessible to others and allowing your contributions to help the world.

It may feel like something completely new if you have not done a scientific report, and that is okay. We will not be doing everything that a manuscript would require but rather looking to gain familiarity with presenting an experimental study. I will talk about the specifics

during the introduction for this lab, and I will provide an instructions sheet to help with the structure of this report.

**Lab Final.** The lab final will test your understanding of the theories utilized in lab sections this quarter as well as the calculations implemented to yield meaningful data. This exam will be during your in-person lab time during the last week of class and is worth **100 points**. You will be allowed to use any notes you have taken throughout the entire quarter during this test, so it is beneficial to organize your work and pay attention during the lab introductions. No early or late exams will be allowed. No working with chemicals required.

## Lecture Schedule

Lecture topics are in black, holidays are in green, and exams are in red. The dates for lecture topics may change but the exam dates will not.

Week Of	Week #	Monday	Wednesday
4/4/2022	1	Flex Day No Lecture	Chapter 13 (Solutions)
4/11/2022	2	Chapter 13 (Solutions)	Chapter 13 (Solutions)
4/18/2022	3	Chapter 19 (Buffers)	Chapter 19 (Buffers)
4/25/2022	4	Exam 1 -Chapter 13	Chapter 19 (Buffers)
5/2/2022	5	Chapter 19 (Buffers)	Chapter 21 (Electrochemistry)
5/9/2022	6	Chapter 21 (Electrochemistry)	Chapter 21 (Electrochemistry)
5/16/2022	7	Exam 2 - Chapter 19	Chapter 21 (Electrochemistry)
5/23/2022	8	Chapter 21 (Electrochemistry)	Chapter 23 (T-Metals)
5/30/2022	9	Memorial Day No Lecture	Exam 3 - Chapter 21
6/6/2022	10	Chapter 23 (T-Metals)	Chapter 23 (T-Metals)
6/13/2022	11	Chapter 24 (Nuclear)	Chapter 24 (Nuclear)

**LECTURE FINAL EXAM: WEDNESDAY June 22, 4:00 PM – 6:00 PM**

## Lab schedule for In-Person Lab Meetings

WEEK OF	WEEK	MONDAY	TUESDAY	WEDNESDAY	THURSDAY
<b>4/4/2018</b>	<b>1</b>	<b>FLEX DAY</b>	<b>FLEX DAY</b>	<b>CHECK-IN (first day)</b>	<b>CHECK-IN (first day)</b>
<b>4/11/2018</b>	<b>2</b>	Freezing Point (1)	Freezing Point (1)	Freezing Point (2)	Freezing Point (2)
<b>4/18/2018</b>	<b>3</b>	Buffers and Titrations (1)	Buffers and Titrations (1)	Buffers and Titrations (2)	Buffers and Titrations (2)
<b>4/25/2018</b>	<b>4</b>	Ksp & Common Ion Effect (1)	Ksp and Common Ion Effect (1)	Ksp & Common Ion Effect (2)	Ksp and Common Ion Effect (2)
<b>5/2/2018</b>	<b>5</b>	Anions (1)	Anions (1)	Anions (2)	Anions (2)
<b>5/9/2018</b>	<b>6</b>	Electrochem (1)	Electrochem (1)	Electrochem (2)	Electrochem (2)
<b>5/16/2018</b>	<b>7</b>	Cations (1)	Cations (1)	Cations (2)	Cations (2)
<b>5/23/2018</b>	<b>8</b>	Cations (3)	Cations (3)	Cations (4)	Cations (4)
<b>5/30/2018</b>	<b>9</b>	<b>MEMORIAL DAY</b>	Cations (5)	Cations (5)	Cations (6)
<b>6/6/2018</b>	<b>10</b>	Cations (6)	Cations (7)	Cations (7)	Cations (8)
<b>6/13/2018</b>	<b>11</b>	Cations (8)	Cations (9)	<b>CHECK-OUT/Lab Final</b>	<b>CHECK-OUT/Lab Final</b>

# Homework

Homework will **NOT** be turned in for credit, but doing all of these is highly recommended for practice and overall concept mastery. These problems are chosen as the minimum needed to practice the topics, but you are strongly encouraged to go beyond the listed problems and try other problems throughout the book. Test questions will be similar to homework questions, so it is important to practice each problem and get help when you need it. I will release an answer key for the problems, but it will correspond to the 9<sup>th</sup> edition only.

Homework Problems for BOTH 8 <sup>th</sup> and 9 <sup>th</sup> Edition	
Chapter 13	4, 6, 7, 24, 26, 29, 30, 31, 33, 35, 37, 41, 46, 51, 57, 60, 65, 67, 70, 75, 84, 86, 94, 97, 101, 103, 111, 120
Chapter 19	4, 7, 11, 13, 17, 25, 27, 31, 38, 40, 51, 53, 62, 71, 72, 74, 75, 78, 80, 87, 88, 90, 91
Chapter 21	10, 13, 17, 24, 30, 32, 34, 39, 41, 43, 47, 52, 59, 63, 65, 69, 71, 88, 90, 92, 101, 102, 105
Chapter 23	4, 5, 11, 15, 21, 31, 36, 46, 48, 54, 56, 72, 73, 78, 81, 88, 89, 91, 93, 96, 97,
Chapter 24	5, 8, 9, 11, 13, 15, 17, 21, 22, 23,

# Lab Safety/Preparedness

Maintaining safety when performing experiments is a primary concern. There are many hazards associated with chemistry labs, so it is essential to recognize these hazards and understand that with proper techniques, the risk drops significantly. There are a few very simple steps students should take to execute safe lab techniques.

First, always wear personal protective equipment (PPE) when performing lab experiments. Such items include, but are not limited to, safety goggles, long pants, sleeved shirts, and closed-toe shoes. **All of this safety equipment must remain on until you complete the experiment, including cleanup.** A detailed list containing safe lab procedures and general practices is given on the next and must be reviewed and signed before starting experiments.

Second, read the lab procedure BEFORE executing the lab procedure. Notes, facts, or some recognition of the hazards is required for the prelab to ensure the section on safety has been read. Reading the procedure ahead of time and knowing what tasks are at hand will also help the experiment go smoothly.

Finally, listen carefully to the directions provided by the instructor. Many techniques can be performed safely and easily with the proper technique but become a safety hazard when performed improperly

What follows is a list from the American Chemical Society Safety In Academic Laboratories Guidelines, 7th Ed., the following mandatory minimum safety requirements must be followed by all students and be rigorously enforced by all Chemistry faculty:

From the American Chemical Society Safety In Academic Laboratories Guidelines, 7th Ed., the following mandatory minimum safety requirements must be followed by all students and be rigorously enforced by all Chemistry faculty:

- 1)** Chemistry Department-approved safety goggles purchased from the De Anza College bookstore (NOT safety glasses) must be worn at all times once laboratory work begins, including when obtaining equipment from the stockroom or removing equipment from student drawers, and may not be removed until all laboratory work has ended and all glassware has been returned to student drawers.
- 2)** Shoes that completely enclose the foot are to be worn at all times; NO sandals, open-toed, or open-topped shoes, or slippers, even with socks on, are to be worn in the lab
- 3)** Shorts, cut-offs, skirts or pants exposing skin above the ankle, and sleeveless tops may not be worn in the lab: ankle-length clothing must be worn at all times
- 4)** Hair reaching the top of the shoulders must be tied back securely
- 5)** Loose clothing must be constrained
- 6)** Wearing "...jewelry such as rings, bracelets, and wristwatches in the laboratory..." should be discouraged to prevent "...chemical seepage in between the jewelry and skin...".
- 7)** Eating, drinking, or applying cosmetics in the laboratory is forbidden at ALL times, including during lab lecture
- 8)** Use of electronic devices requiring headphones in the laboratory is prohibited at ALL times, including during lab lecture
- 9)** Students are advised to inform their instructor about any pre-existing medical conditions, such as pregnancy, epilepsy, or diabetes, that they have that might affect their performance.
- 10)** Students are required to know the locations of the eyewash stations, emergency shower, and all exits
- 11)** Students may not be in the lab without an instructor being present
- 12)** Students not enrolled in the laboratory class may not be in the lab at any time after the first lab period of each quarter.
- 13)** Except for soapy or clear rinse water from washing glassware, NO CHEMICALS MAY BE Poured INTO THE SINKS; all remaining chemicals from an experiment must be poured into the waste bottle provided.
- 14)** Students are required to follow the De Anza College Code of Conduct at all times while in lab: "horseplay", yelling, offensive language, or any behavior that could startle or frighten another student is not allowed during lab;
- 15)** Strongly recommended: Wear Nitrile gloves while performing lab work; wear a chemically resistant lab coat or lab apron; wear shoes made of leather or polymeric leather substitute.

By signing below, I, \_\_\_\_\_,  
First Name Family Name

acknowledge that I fully understand and agree to abide by the laboratory safety rules listed above. Further, I acknowledge that my failure to abide by these rules will result in my being dropped from this chemistry class immediately.

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

**Student Learning Outcome(s):**

\*Apply the principles of equilibrium and thermodynamics to electrochemical systems.

\*Apply the principles of transition metal chemistry to predict outcomes of chemical reactions and physical properties.

\*Evaluate isotopic decay pathways.

\*Demonstrate a knowledge of intermolecular forces.