Quantum Mechanics in 1D

- 1. A matter wave called the wavefunction Ψ can be associated with any particle.
- 2. The wavefunction Ψ contains all the information that can be known about the particle,

These 2 statements bring up two important questions:

- 1. How do we obtain this wavefunction Ψ for a given system?
- 2. How do we extract information from this wavefunction?

Let's look at the answer to the 2nd question first.

Born Interpretation of Ψ

The probability that a particle will be found in the infinitesimal interval dx about the point x, denoted by P(x)dx is given by:

$$P(x)dx = |\Psi(x,t)|^2 dx$$

 $P(x) = |\Psi(x,t)|^2$ (probability density function)

P(x) = probability per unit length (in 1D)

- A. It follow from this definition that you cannot specify with certainty the location of a particle! You can only specify the probability!
- B. $\Psi(x,t)$ itself is NOT a measurable quantity but $|\Psi(x,t)|^2$ is measurable and equal to the probability per unit length (probability density) of finding the particle in the interval dx about the point x.
- C. $P(x) = |\Psi(x,t)|^2 = \Psi \Psi^*$ where Ψ^* is the complex conjugate of Ψ .
- D. $\Psi(x,t)$ must be single-valued and continuous function of x and t.
- E. Because the particle must be somewhere along the x-axis, the sum of the probabilities over all values of x must add up to 1:

$$\int_{-\infty}^{+\infty} P(x) dx = \int_{-\infty}^{+\infty} |\Psi(x,t)|^2 dx = 1$$
 Normalization Condition

F. The probability of finding the particle in any finite interval $a \le x \le b$ is given by:

$$P_{ab} = \int_{a}^{b} |\Psi(\mathbf{x}, \mathbf{t})|^{2} dx$$

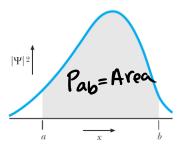


Figure 6.1 The probability for a particle to be in the interval $a \le x \le b$ is the area under the curve from a to b of the probability density function $|\Psi(x, t)|^2$.

G. $|\Psi(x,t)|^2 \to 0$ fast enough as $x \to \pm \infty$ so that the normalization condition holds valid.